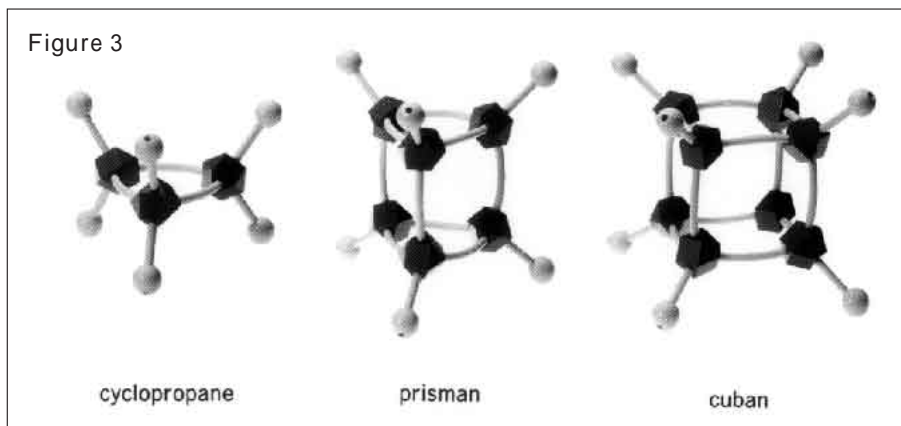
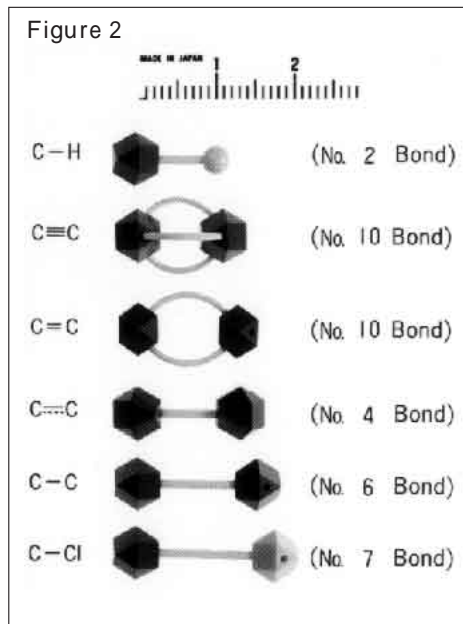


Bond distance has been designed at 1 = 2.5cm and when, firmly inserting bonds into holes drilled on balls, distance between the centers of balls is measured with the centers of balls is measured with the attached rule, an inter-ball distance can be directly read in the unit of (Figure 2).

Because of moderate elasticity of the bonds, it is easy to assemble deformed compounds. It requires special parts in the case of conventional metallic models (Figure 3).



- ④ Because bonds are rotated smoothly and furthermore, can be fixed at any position, configurations and their changes can be observed without special stoppers, which are needed on metallic models (Figure 4).
- ⑤ Atoms available in various colors make it possible to learn the basic stereochemistry using Atoms in any color in addition to the standard usage (Figure 5).
- ⑥ It is possible to learn the principle of high-level organic reactions such as the Woodward-Hoffman rule using a *p* orbital plates (Figures 1 and 6)